

# Validating the Oxygen Performance of a Wine Closure

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# ABSTRACT

Oxygen plays a key role in wine chemistry and aging as it affects the stability and sensory properties of wine. Amongst the current choices of wine closures the discussion of oxygen permeability has come to the forefront, but there is little agreement as to what methods are appropriate for measuring oxygen transmission rates (OTR) in a wine context or even what the target OTR should be. Given the high number of reported faults in bottled wines stemming from over, or under oxidation, the ability to measure oxygen transmission through a closure and to insure that a given closure is delivering the intended oxygen dose is of significant commercial importance for the wine industry as a whole. Our research has two main objectives. 1. experimentally test our theories about optimal wine oxidation levels. 2. Confirm that our closures perform within that range.

To accomplish these objectives, wine enclosures are tested with model wines bottled in clear glass where oxygen ingress is measured by different oxygen testing methods, and a variety of commercial wines fitted with closures of known oxygen permeability characteristics, which can be tested for post-bottling SO<sub>2</sub> decline and other chemical parameters.

We will present (i) the **theory** behind our targeted level of in-bottle wine oxygenation, (ii) our **material and methods** for validating it, and (iii) the **results of our research to-date** in creating a closure that reliably delivers oxygen within that range.

# INTRODUCTION – Theory behind VinPerfect's targeted OTR range

How do we know how much oxygen a wine needs? It is important to understand that trying to have one "optimal" amount of oxygen is not realistic. However, it is possible to define a range of oxygen ingress rates that might be advantageous to wine quality. The approach is to consider multiple products that address different rates of oxygen ingress and then further let the winemaker decide what is right for the wine.

Looking at the current closure market there is no product that consistently performs within that range we target. Synthetic corks admit too much oxygen, screw caps admit too little oxygen, and natural cork is very inconsistent (see Figure 1.) (5, 9)

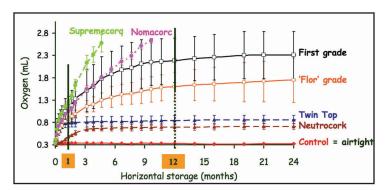


Figure 1. Oxygen ingress true closures (9).

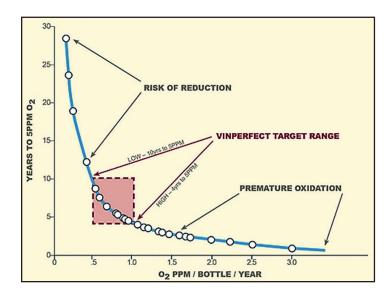
VinPerfect's defined ranges are associated to the problems a wine will bear post-bottling, too much and a wine might prematurely oxidize, lose varietal characteristics, decrease in pleasant aromatic compounds and increase in aldehydes (1, 3, 6, 11, 12). Too little oxygen and a wine might under develop and is in danger of forming reduced aromas (3, 7, 8, 10, 13).

Further considerations are that the customer does not see wine as a perishable product. How long should a wine last before becoming oxidized? For most wines, a storage of below a year is adequate, but it should be possible to store a wine at least five years before becoming oxidized.

### How do we define shelf life of a wine?

- (1) Wine is protected against oxidation products by free SO<sub>2</sub>. We take the mass ratio of SO<sub>2</sub> and O<sub>2</sub> into consideration ( $2 \text{ SO}_2 + \text{O}_2 \rightarrow 2 \text{ SO}_3$ ), where 1 mg/L of O<sub>2</sub> will consume 4 mg/L of SO<sub>2</sub>(2, 5)
- (2) Most wines are bottled at 30 mg/L free SO<sub>2</sub>
- (3) When free SO<sub>2</sub> drops below 10 mg/L wines' capacity of binding aldehydes is significantly lowered and oxidations products start to appear (2).
- (4) As a consequence of points (1) (3), and considering the "lightest" kind of wine, e.g. a white wine, following oxygen transmission for our upper and lower limits can be extracted:

For a wine decline in free  $SO_2$  from 30 mg/L to 10 mg/L, the minimum  $O_2$  required is 5 mg/L, which gives us a shelf life of 4 years and an  $O_2$  annual ingress of 1 mg/L. It is harder to define the lower limit clearly, but with an annual ingress of 0.5 mg/L, a shelf life of 10 years is established, which insures against reduction problems (Figure 2).





# How does the winemaker choose an oxygen level within this range?

If we consider the light white wine example, which is the most vulnerable to oxidation, a winemaker should probably utilize a closure closer to the low-oxygen end of the range at 0.5 ppm per year. Conversely, a red wine with elevated phenolic content and therefore a higher oxygen absorption capacity could benefit more from an oxygen rate close to 1 ppm per year. A red wine has less to risk from the upper oxygen limit.

In the end, the selection of a closure's oxygen transfer rate for a wine's post-bottling life is an extension of the winemaker's style. So, we would not consider to make specific suggestions of oxygen transfer rates for various varieties.

In summary for deliberation, for an age worthy red wine 1 ppm per year will allow gradual development in the bottle for 10 to 20 years, without being starved of oxygen or over-exposed. For white wines, the selection of a low oxygen closure means that even the lightest white may last 10 years or more. The range we have selected is right in the center of the average cork oxygen performance range when the bottle is stored properly but without its inconsistency (Figure 1).

# MATERIALS AND METHODS

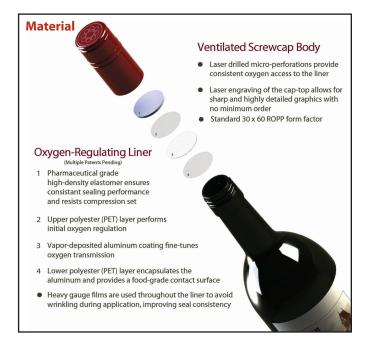


Figure 3. VinPerfect's Closure System

### Method for Ingress Determination:

Wine bottles were filled with deoxygenated model wine solution (12% ethanol, distilled water, tartaric acid, adjusted to pH 3.6). Non-invasive oxygen sensors from NOMASense and MOCON OpTech (see figure 4, 5 and 6). are placed mid-way and in the headspace the same 375 mL wine bottles. Wine bottles are inverted, stored in a temperature controlled cabinet at 20° and measured weekly for  $O_2$  ingress. Total packaged oxygen is calculated and expressed as mg/L.



Figure 4. NOMASense  $O_2$  Trace (Presens-Fibox 3LCD trace v 7 oxygen anlayzer using Pst6 oxygen sensing ports (range 0–1.8 mg/L and 0–4.2%  $O_2$ ).

Calibration is performed by using a provided factory calibration sheet specific to the Pst6 sensors used.

# a.

Figure 5. a) MOCON OpTech-O<sub>2</sub> Analyzer and b) Calibration Card



Calibration is performed with a MOCON OpTech-O<sub>2</sub> CalCard (see Figure 2b)



Figure 6. Non-invasive oxygen sensors MOCON OpTech and NOMASense side by side in solution and in the headspace

# **ACKNOWLEDGEMENTS**

**RESULTS OF RESEARCH TO-DATE** 3.0 2.5 2.0

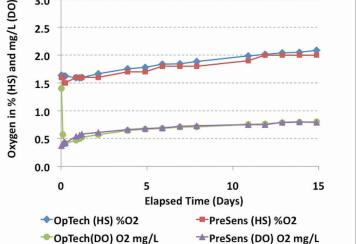


Figure 7. Comparison Trial OpTech O<sub>2</sub> Analyzer versus NOMASense O<sub>2</sub> Trace measuring head space (HS) and dissolved oxygen (DO) over time in wine.

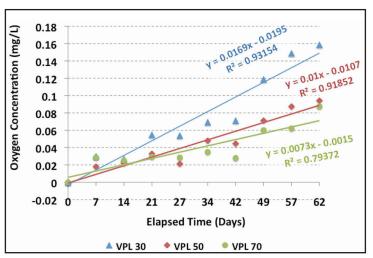


Figure 8. Three levels of liners metalization.

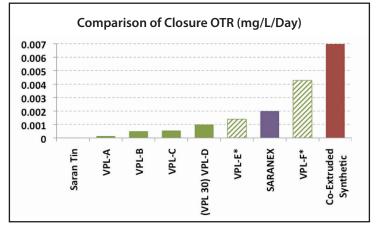


Figure 9. VinPerfect comparison of closure OTR (mg/L/day) and prediction for \*New Liner Material.

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# **CONCLUSIONS**

- > Both NOMASense and MOCON OpTech oxygen analyzers show almost identical performance in measuring head space and dissolved oxygen and work excellent and a winelike-solution. The MOCON instrument has the advantage of being more cost efficient for sensors and the instrument itself. MOCON OpTech is also more user-friendly, and includes software to record and track data easily.
- > There is an inverse correlation between the optical density of the films used in the VinPerfect prototypes and the OTR of the resulting closure. The higher the optical density, the lower the OTR. This indicates a method of positive control of oxygen through the varying of metalization level.
- > Preliminary results show low coefficients of variation in O<sub>2</sub> ingress for VinPerfect's liner prototypes indicating consistency.
- > Measurements of inserted closures with high initial DO did not render useful results with this method since the rising O<sub>2</sub> started to approach equilibrium with the outside environment, reducing the driving force O<sub>2</sub> transmission. These closures should be tested with a larger volume bottle, or better yet, closures should be depleted of contained O<sub>2</sub> by storing them before the test in an  $O_2$  void environment. Further, the first two weeks after bottling should be kept in an inert environment to complete the recovery phase without atmospheric O<sub>2</sub> interference and to keep the equilibrium driving force equal to real-world conditions.
- > Some screw capped samples across manufacturers were shown to have variants not explained by the composition of the liner. In these cases it was found that defects in the sealing-surface of the bottle were present, highlighting the fact that the cap liner is only half of the seal, and underscoring the importance of the vigilance in glass quality.
- > Further research needs to be performed:
  - Trials with inserted closures on larger bottles
  - Correlation of OTR measured via this method, with SO<sub>2</sub> changes observed in real wines
  - Correlation of wine composition, color and sensory to differences in the SO<sub>2</sub> depletion of wines with closures of known OTR.

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